

# UAN SOURCES TODAY

Michael Orr  
Specialty Process Consulting

Kansas

12-10-2014 Sacramento, CA



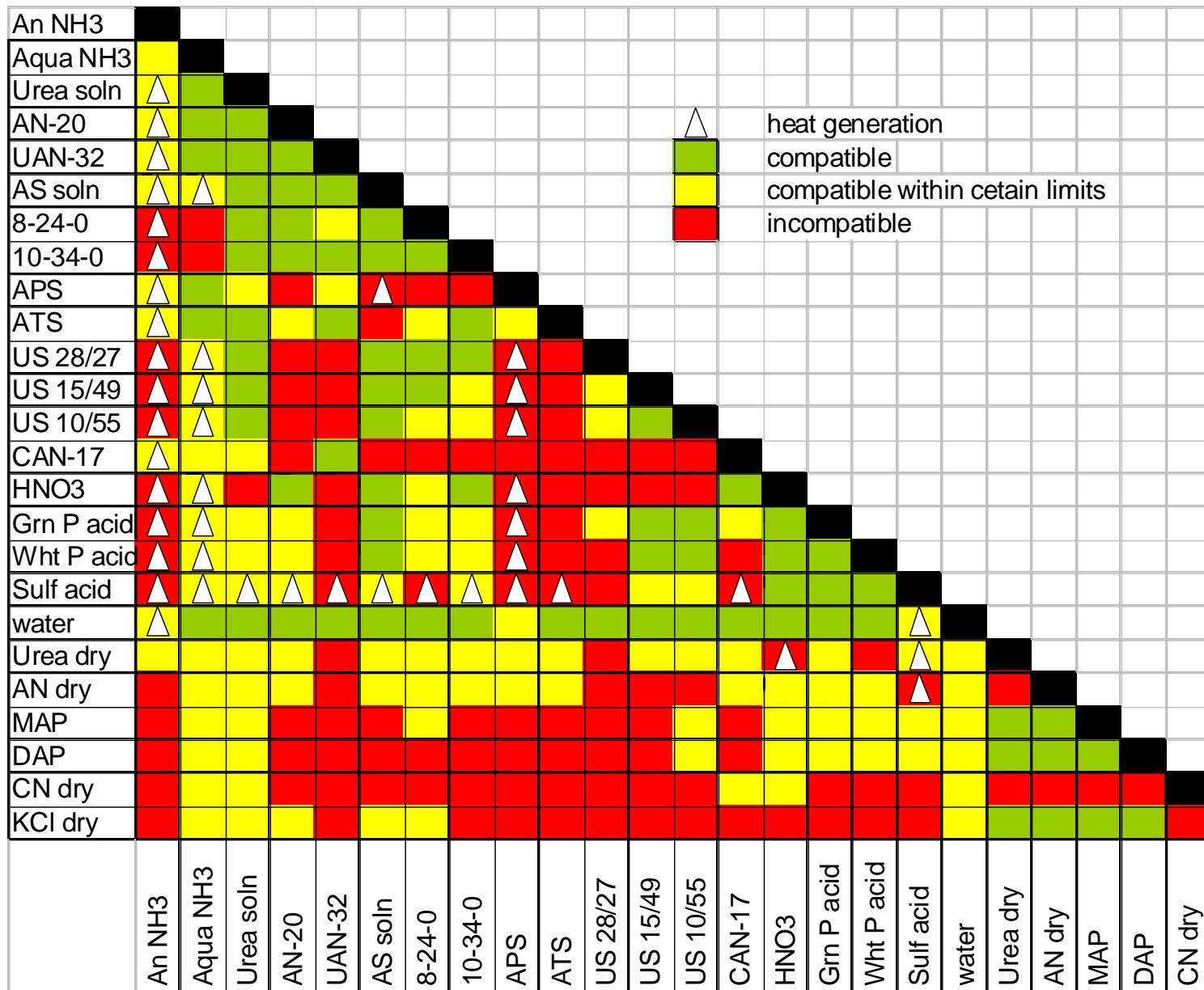
VINCITA

25  
34  
33  
32  
31  
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28  
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26  
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24





# Unocal Compatibility Chart





# UAN PRODUCTION

- AN Liquor + Urea Liquor
- Nitric acid + Ammonia + Urea Liquor
- Melt
- Adjust concentration

# % N salts at Temps

Temp	urea	$\text{NH}_4\text{NO}_3$	UAN	$\text{NH}_4\text{SO}_4$		$\text{NH}_4\text{Cl}$	
32	18.7	19.0		32.6	8.8	10	5.9
35	19.2	19.4		33.6			
40	19.9	20.2		34.8			
45	20.6	20.8		35.7			
50	21.3	21.4		36.3	8.8	10.2	6.5
55	22.1	22.0		36.8			
60	22.8	22.4		37.2			
65	23.5	22.9		37.6			
70	24.2	23.4		37.9	9.1	10.4	7.2
75	25.0	23.8		38.1			
80	25.8	24.3		38.3			
85	26.5	24.7		38.5	9.3	10.6	7.6

# UAN 28-32%

- RATIOS VARY-INFLUENCE BLENDS
- pH CHANGES DUE TO UREA/NH<sub>3</sub>/NO<sub>3</sub>
- ADDITION OF OILS/INHIBITORS
- CROSS CONTAMINATION
- WINTER STORAGE INFLUENCE
- BLENDING

# ADJUST UAN

- Addition of Urea-summer, winter blend
- Addition of Ammonia to adjust pH
- Addition of one or the other also impacts saltout temperature
- Addition of water to cut concentration impacts saltout
- Inhibitor addition impacts saltout temperature

pH	Free NH <sub>3</sub>
0 - 6.30	0.00
6.31 - 6.85	0.01
6.86 - 7.10	0.02
7.11 - 7.25	0.03
7.26 - 7.35	0.04
7.36 - 7.45	0.05
7.46 - 7.52	0.06
7.53 - 7.57	0.07
7.58 - 7.61	0.08
7.62 - 7.67	0.09
7.68 - 7.72	0.10
7.73 - 7.75	0.11
7.76 - 7.80	0.12
7.81 - 7.83	0.13
7.84 - 7.86	0.14
7.87 - 7.88	0.15
OVER 7.88	TITRATE

Sample Name: UAN 32

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Analysis Name	Analysis	Units
Free Ammonia	0.03	%
Ammonium Nitrate	46.52	%
Ammonium Nitrate (N)	16.28	%
Urea	34.32	%
Urea Nitrogen (N)	15.99	%
Total Nitrogen (N)	32.29	%
Biuret	1.0	%
pH	7.53	
Specific Gravity @ Ambient/ 60 deg F	1.3319	

Sample Name: UAN 32

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Analysis Name	Analysis	Units
Free Ammonia	0.03	%
Ammonium Nitrate	45.79	%
Ammonium Nitrate (N)	16.03	%
Urea	35.19	%
Urea Nitrogen (N)	16.40	%
Total Nitrogen (N)	32.44	%
pH	7.57	
Specific Gravity @ Ambient/ 60 deg F	1.3308	

# CHANGING NITROGEN SUPPLY

- Ammonium nitrate
- Increased importation of UAN
- pH variation in UAN
- AN/Urea ratio
- Supply
- Discoloration

# NITRATE/ NO<sub>X</sub>

NO WITH ACIDS

SIGNS REDDISH/BROWN GAS

NERVE GAS/DEADLY

NITRATE NOT HIGHLY SOLUBLE WITH POTASSIUM

WILL COMPETE WITH OTHER ANIONS IN SOLUTIONS

KEEP IN MIND WITH UAN SOLUTION MIXING

# UAN/10-34-0 blend





# Clear Liquid Formulations

Some Keys to Avoiding Salt-out  
Problems

# Salt-out information

- Salt-out results from the formation of insoluble material in a mixed, liquid fertilizer
- It usually occurs between the time the mixture was produced and the time it is actually applied to the field
- In most cases, this decrease in solubility is due to a drop in temperature
- The material which “salts out” may or may not be one of the original ingredients

**SOLUBILITIES OF SALTS IN POUNDS PER 1000 POUNDS OF  
WATER AT DIFFERENT TEMPERATURES**

TEMPERATURE	AMMONIUM NITRATE	UREA	NITROGEN SOLUTION 32-0-0	POTASSIUM CHLORIDE
FO				
32	1180	670	3900	280
35	1246	698	4552	285
40	1357	745	5639	293
45	1479	793	6725	302
50	1580	840	7812	310
55	1683	898	8900	318
60	1786	956	9987	327

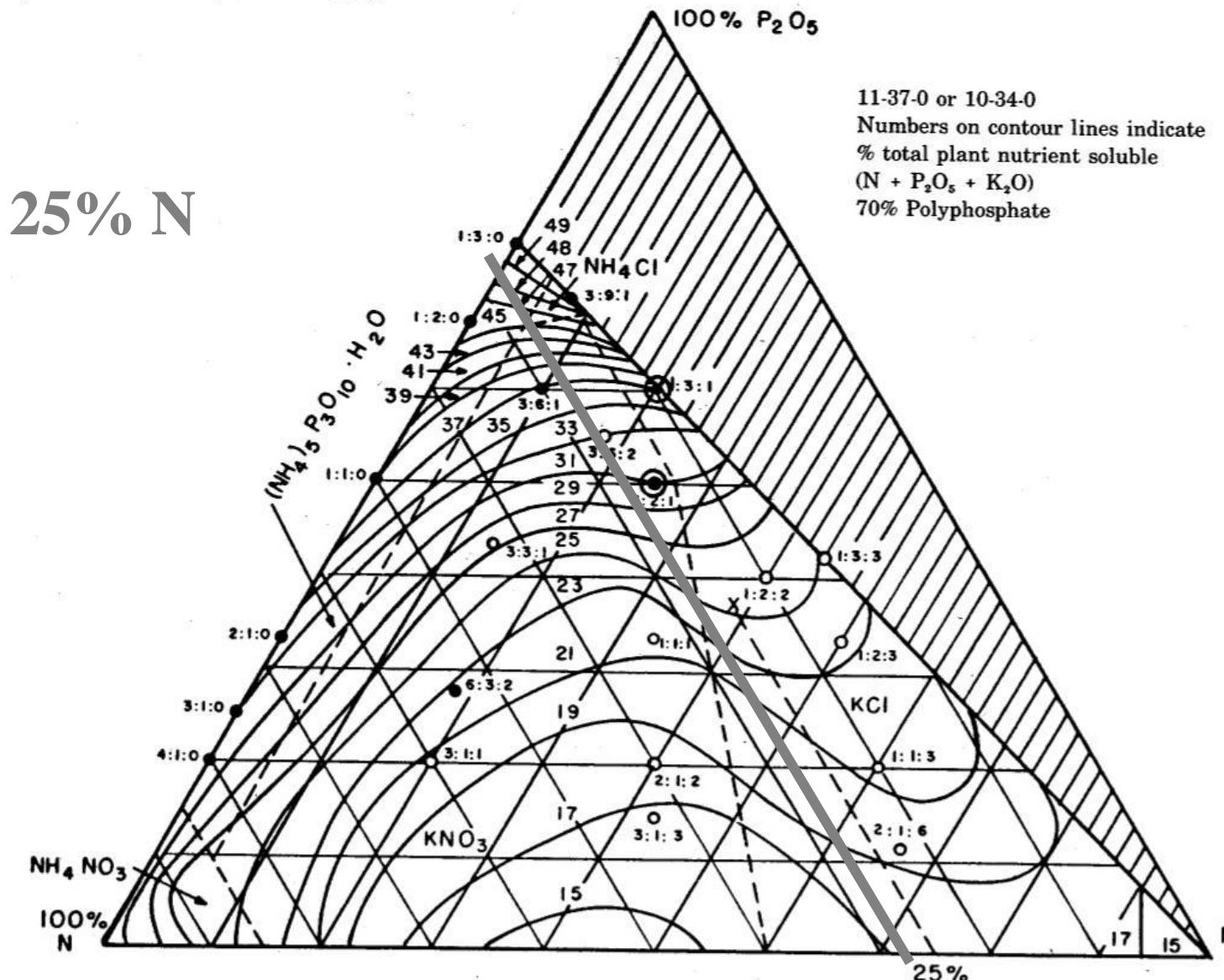
# Where to start?

- Salt-out tables (the no brainer approach)
  - Exist for both 32°F and 50°F Salt-out T's
  - Choose UAN or Urea
  - Choose 0%, 45% or 70% Poly
- Triangular diagrams
  - Each set of raw materials has its own chart
  - Most charts are for 32°F salt-out
  - Most are for either 0% or 70% Poly

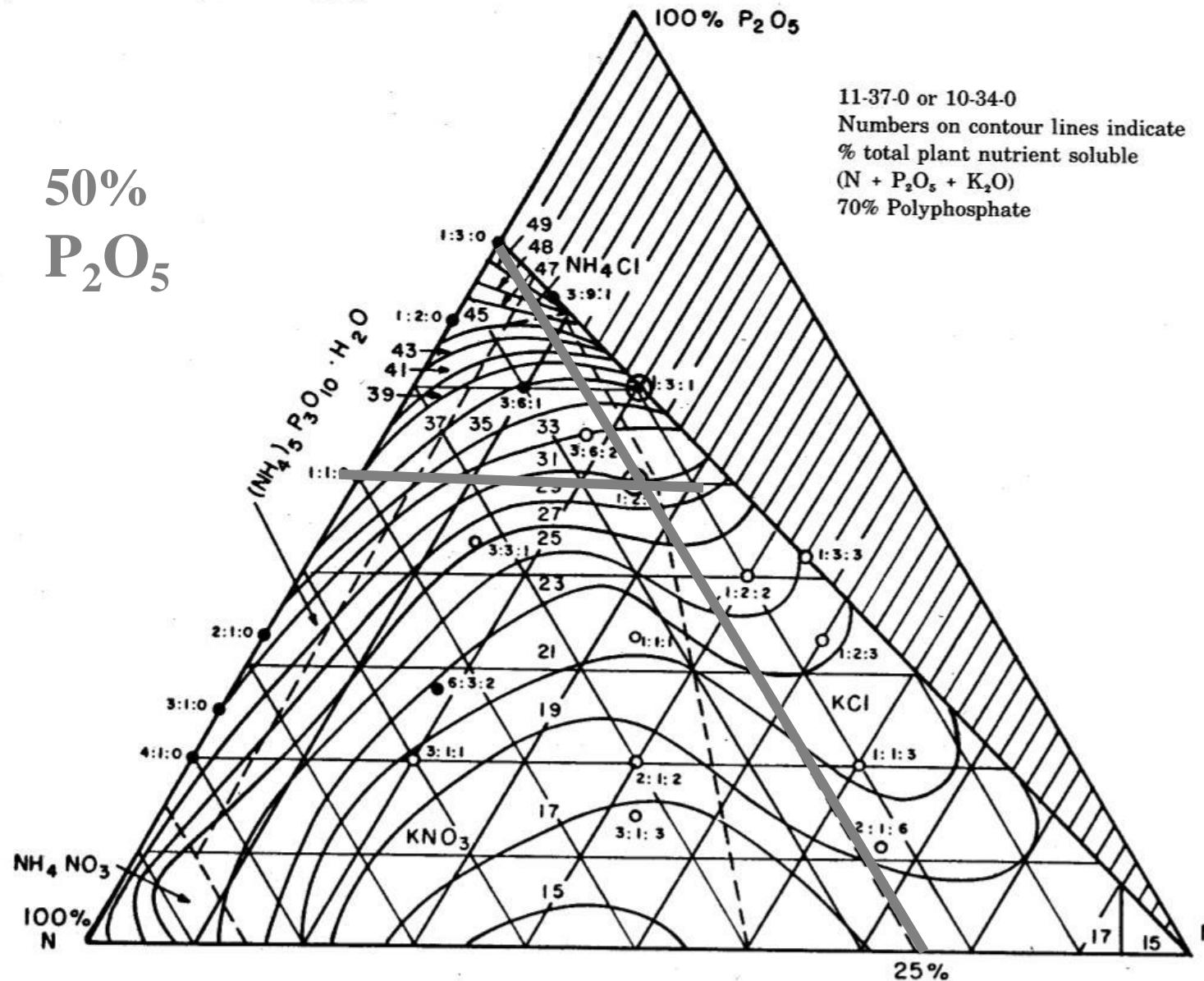
# Triangular Charts (Cont'd)

- Advantages over tables
  - Show total plant food units possible for a given grade ratio
  - Indicate what material will salt-out first (KCl, NH<sub>4</sub>, NO<sub>3</sub>, etc.)
- Drawbacks
  - Most are for 32°F only
  - A little harder to read (but not much)

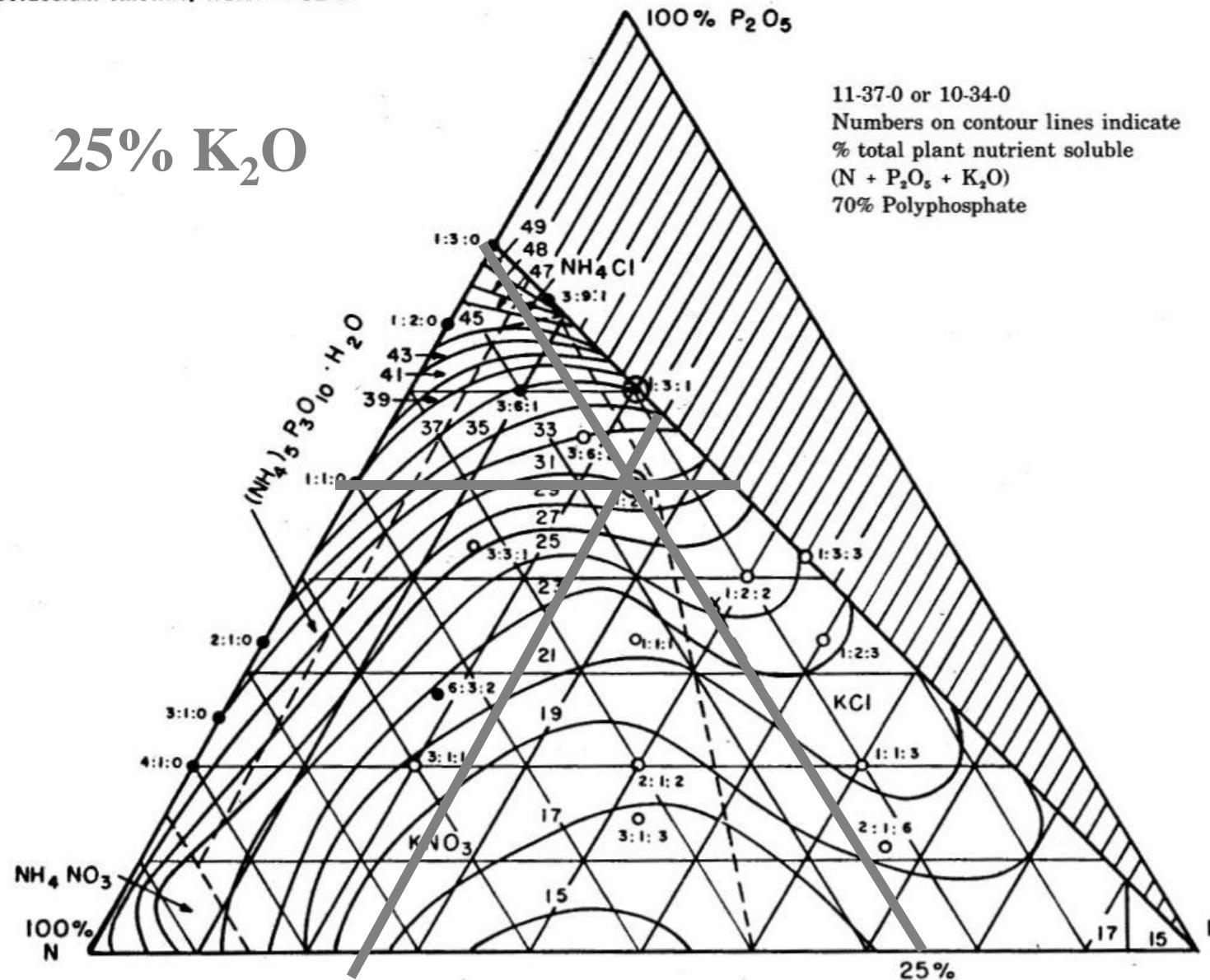
**Figure 3.7: System 11-37-0, UAN solution,  
potassium chloride, water at 32°F.**



**Figure 3.7: System 11-37-0, UAN solution,  
potassium chloride, water at 32°F.**



**Figure 3.7: System 11-37-0, UAN solution,  
potassium chloride, water at 32°F.**



# Question – if the temperature falls below 32°F, what material will precipitate?

- If you have a Potash/UAN Mixture
- Answer –  $\text{KNO}_3$ /

The area inside the dashed lines defines what material will precipitate







**Intertek Caleb Brett**

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Intertek Caleb Brett Ref. #:

US621-12095

**CERTIFICATE OF QUALITY**  
(ISSUED AT THE DISCHARGE PORT)

An Inspector from our New Orleans, LA branch sampled the herein described shipment prior to discharge from the vessel. The sample was analyzed at the Intertek, New Orleans laboratory with results as follows:

**VESSEL :** M/T STREAM NEPTUNE  
**DESCRIPTION OF GOODS :** UREA AMMONIUM NITRATE, 32 % N  
**TERMINAL :** BURNSIDE ANCHORAGE  
**EX VESSEL'S TANK :** 1P, 1S, 3P, 3S, 5P, 5S, 8P, 8S, 9P & 9S  
**DISCHARGED TO BARGES(S) No. :** KIRBY 30310T, KIRBY 27304, KIRBY 29001, KIRBY 29306, KIRBY 28001T,  
KIRBY 18500B, KIRBY 29003, KIRBY 20500B, KIRBY 30320, KIRBY 20504,  
KIRBY 27000, KIRBY 28100B  
**SHIPMENT TONNAGE:** 32,827.329 METRIC TONS  
**DISCHARGE DATE (S) :** AUGUST 24, 2014

SPECIFICATIONS:	TEST METHOD	SPECIFICATIONS	RESULTS
Appearance:	VISUAL	Translucent, slightly cloudy liquid	Transparent slightly cloudy liquid
Total Nitrogen Content, %	ADAC 903.13	32.0 MTN	32.42
Molar Ratio Urea / Ammonium Nitrate	CALCULATION	0.74-0.80	0.81
Alkalinity In terms of free Ammonia, %	T.F.I. 11A.7	0.30 MAX	0.03
Density @ 15 deg. C @/CM3 (°)	ASTM D4032	None Specified	1.32886
Density @ 15 deg. C @/CM3 in Air	By conversion	-	1.32775
Density @ 40 deg. C @/CM3	ASTM D4032	1.293-1.326	1.31123
Nitrogen in the form of Ammonia & Nitrate, %	T.F.I. 11A.6	None Specified	15.58
Nitrogen in the form of Urea, %	T.F.I. 11A.4 / By Difference	None Specified	16.84
Fraction of Total Mass of Nitribor, ppm	L.WL.027.L	None Specified	<0.001
Solidification Temp, Deg. C	L.WL.024.L	-2 MAX	-2

Upon arrival at the discharge port on 8/2/2014, samples were drawn from all above-mentioned Vessel tanks and a composite sample was prepared. This analysis conducted on the Vessel composite sample prior to discharge, represents the quality of the total cargo on board and not for individual parcels (if any) subsequently discharged.

For and on behalf of  
Intertek Agricultural Services

Nikki Helbert, Operations Manager

Signed and dated in: St. Rose, LA  
on: August 6, 2014



INSPECTORATE

## ANALYSIS CERTIFICATE

RT11706

This is to certify that Inspectorate Estonia AS, at the request of "██████████", performed inspection of ships tanks of the under mentioned vessel:

Commodity	UREA AMMONIUM NITRATE SOLUTION (UAN) 32% BULK
Loaded into the	M/T "BOW FORTUNE"
Place of loading	SILLAMAE, ESTONIA
Date of loading	JULY 7 <sup>th</sup> -9 <sup>th</sup> 2014
Loaded quantity as per BVL no.1 dated 09/July/2014	29 156.591 MT
Destination	1-2 SB CORPUS CHRISTY, TX U.S.A.

RESULTS OF CHEMICAL ANALYSIS

Sampling and chemical analysis based on composite sample taken from vessel's tanks after loading was performed according to the requirements and the following results were obtained:

Tests	Methods	Units	Specifications	Results
Appearance	VISUAL	Result	The Tinglic Pigment Clear	TCLP
Total Nitrogen Content	AOAC 993.13	%mass	Liquid min 32.0	32.03
Urea/Amonium Nitrate	CALCULATED	%mass	0.74-0.80	0.79
Alkalinity in terms of free Ammonia	ADCH0561 (BORDEN)	%mass	0.02-0.30	0.083
Crystallization temperature, °C	ASTM D 2500 (mod.)	°C	Not higher than -2°C	-14
Content of Inhibitor (CORBLOCK 9140 or identical)	ACRON in-house method	%mass	0.01 min	0.011
Density of liquid @40°C,	ASTM D 4052 (mod.)	g/cm3	1.293-1.326	1.3214



Dated: 09th July, 2014  
For and on behalf of  
**INSPECTORATE AS**

Inspectorate Estonia AS  
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Registered in Estonia no. 101 55092  
Certified on ISO 17020:2006  
EAC 17/35 accredited Certificate L409  
EVS-RN-4501 accredited Certificate P0014

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## CERTIFICATE OF QUALITY

5089146

This is to certify that, at the request of Messrs. ██████████, we have performed the quality inspection of the under-mentioned goods:

Description of goods: **UAN SOLUTION, MINIMUM 32% NITROGEN**  
Vessel: **MT FAIRCHEM COLT**  
Port of loading: **CONSTANZA, ROMANIA**  
Date of loading: **08<sup>TH</sup> - 10<sup>TH</sup> OF NOVEMBER, 2012**  
Port of discharge: **1-3 PORT(S) USEC (PLILA-JAX RGE) OR 1 SP USG  
RANGE INCL. MISS. RIVER (NOT NORTH OF D'VILLE), USA**  
Quantity as per B/L.  
No. 1 dated 30.11.2012 **18,450.000 MT**

### RESULTS OF INSPECTION

We hereby certify that we took samples from ship's tanks after loading of **UAN SOLUTION, MINIMUM 32% NITROGEN** and, based on analysis of the ship's composite sample, report the following findings:

Parameters	Specifications	Methods	Results
Total Nitrogen	32 % min.	ISO 2003/2006 method 2.6.2	32.09 %
Aminic Nitrogen	16.0 % +/- 1.1	SR ISO 5661:1993 SR EN ISO 12185:2002	15.99 %
Nitrate Nitrogen	16.0 % +/- 1.1	ISO 2003/2006 method 2.1 / 2.2.3	16.10 %
Corrosion inhibitor (SK FERT UAN)	180 ppm max	SNM - SLUISKU-	160 ppm
Crystallization point	< or = 0°C	SR EN ISO 1392:1994	0°C
Pressure at 40°C	0	Manometer tube U with water	29
Free Ammonia	0.1% max	SR ISO 1593:1999	0.04 %
Specific weight at 20°C	1.320 g/cm <sup>3</sup> min.	ISO 758:1976 SR EN ISO 12185:2002	1.3261 g/cm <sup>3</sup>
Aspect	Clear liquid	Visual	Clear liquid

Dated: 18<sup>th</sup> of November, 2012

For and on behalf of:

INSPECTORATE (SUISSE) S.A.

F40491



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Intertek Ref. #:

U9620-00001918

**CERTIFICATE OF QUALITY**  
(ISSUED AT THE DISCHARGE PORT)

An Inspector from our Wilmington, NC branch sampled the herein described shipment prior to discharge from the vessel. The sample was analyzed at the Intertek, New Orleans laboratory with results as follows:

**VESSEL :** MT "FAIRCHEM COLT"  
**DESCRIPTION OF GOODS :** UAN SOLUTION 32 PCT NITROGEN IN BULK  
**TERMINAL :** ██████████ - WILMINGTON, NC  
**EX VESSEL'S TANK :** 1P, 1S, 2P, 2S, 3P, 3S, 7P, 7S, 8P, 8S, 10P & 10S  
**DISCHARGED TO SHORE TANK(S) No. :** SOUTH  
**SHIPMENT TONNAGE:** 18,468.574 METRIC TONS  
**RECEIVER :** ██████████  
**DISCHARGE DATE (S) :** DECEMBER 01-02, 2012  
**DATE(S) TESTED:** DECEMBER 04, 2012

**SPECIFICATIONS:**

**TEST METHOD**

**SPECIFICATIONS**

**RESULTS**

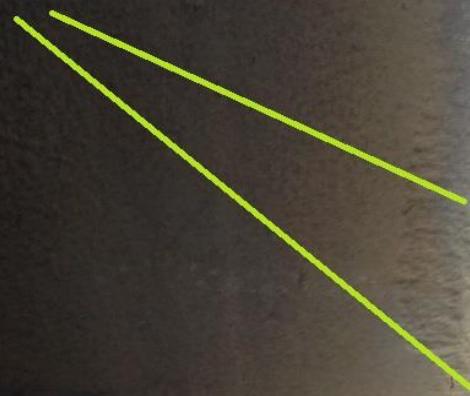
Appearance:	Visual	Transparent slightly dyed liquid	Transparent slightly dyed liquid
Total Nitrogen Content, %	ADAC 963.13	32.0 MIN	32.08
Mass Ratio-Urea / Ammonium Nitrate	CALCULATION	0.78 +/- 0.05	0.72
Alkalinity in terms of free Ammonia, %	T.F.I. 11.A.7	0.01 - 0.5	0.04
Density @ 15 deg. C GCM 3 (%)	ASTM D4052	None Specified	1.3294
Density @ 15 deg. C GCM 3 in Air	By conversion	-	1.3284
Nitrogen in the form of Ammonia & Nitrate, %	T.F.I. 11.A.6.	14-18	8.16
Nitrogen in the form of Urea, %	L.L. 11.A.14/ By difference	14-18	15.73
Fraction of Total Mass of Inhibitors as P2O5, ppm	LWI 027.L	None Specified	3.6

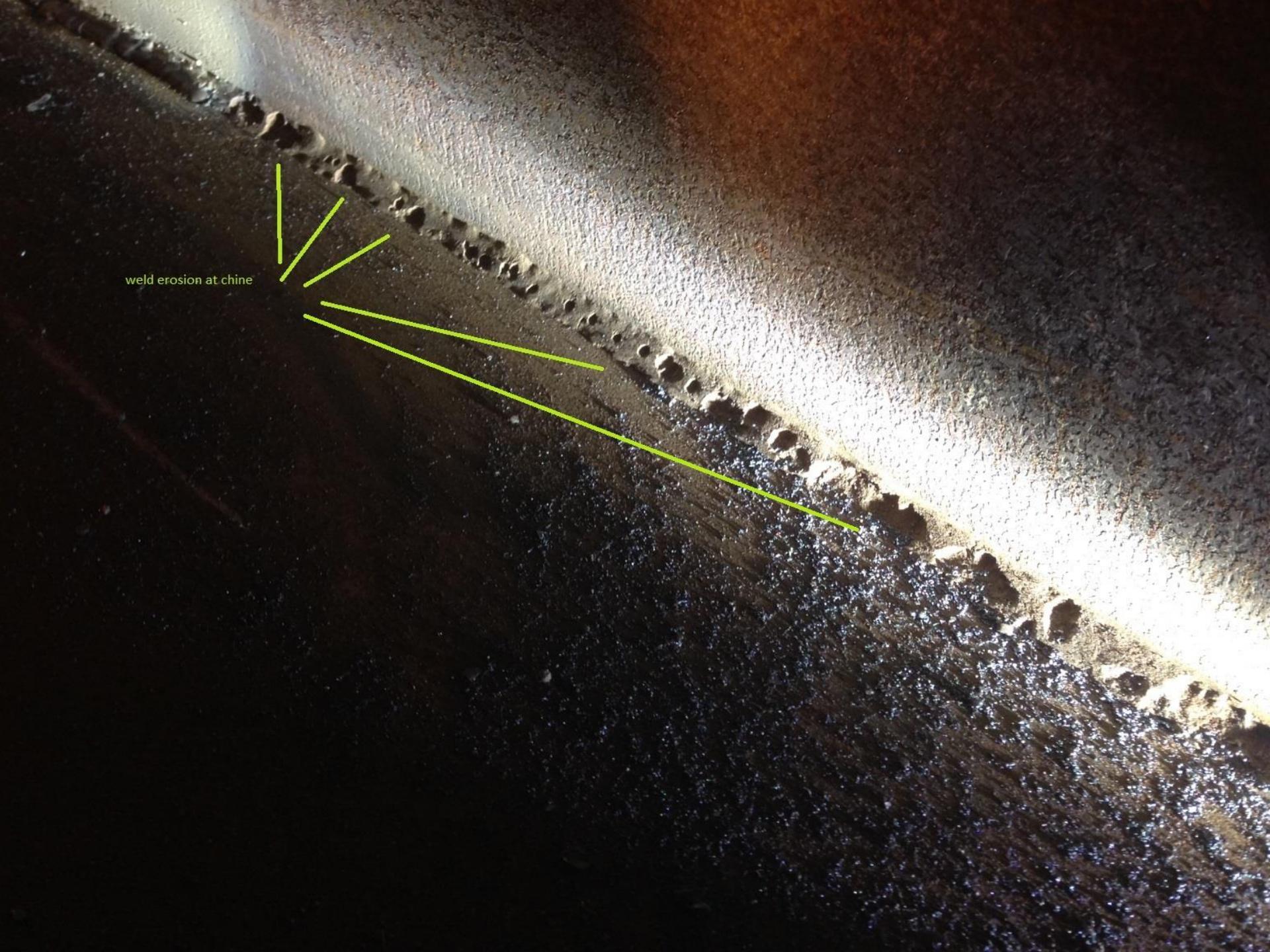
(\*) Tested at Intertek, Wilmington Lab

Upon arrival at the first discharge port on 12/01/2012, samples were drawn from all above-mentioned Vessel tanks and a composite sample was prepared. This analysis conducted on the Vessel composite sample prior to discharge, represents the quality of the total cargo on board and not for individual parcels (if any) subsequently discharged.

Intertek

weld erosion at bottom  
of tub ring vert





weld erosion at chine

This image shows the underwater hull of a vessel. The hull is made of metal plates, and a prominent vertical seam or chine is visible along the side. The metal surface appears dark and textured. A series of bright, glowing areas indicate where the hull has been eroded, particularly along the chine. Three yellow lines have been drawn to highlight these eroded areas. The first yellow line is a short vertical segment pointing to a bright spot near the top of the chine. The second yellow line is a diagonal segment starting from the same area and extending downwards and to the right. The third yellow line is a longer diagonal segment starting from the second one and extending further down the chine. The text "weld erosion at chine" is written in white in the upper left quadrant of the image, identifying the specific location of the damage.

# THE FUTURE

- Know what the AN/UREA ratio
- Consider pH (ammonia) above pH 7
- Know total Nitrogen
- Inhibitor?
- A Crystal Ball would be nice

